

Denitrification in Suburban Lawn Soils

Steve M. Raciti,* Amy J. Burgin, Peter M. Groffman, David N. Lewis, and Timothy J. Fahey

There is great uncertainty about the fate of nitrogen (N) added to urban and suburban lawns. We used direct flux and in situ chamber methods to measure N_2 and N_2O fluxes from lawns instrumented with soil O_2 sensors. We hypothesized that soil O_2 , moisture, and available NO_3^- were the most important controls on denitrification and that N_2 and N_2O fluxes would be high following fertilizer addition and precipitation events. While our results support these hypotheses, the thresholds of soil O_2 , moisture, and NO_3^- availability required to see significant N_2 fluxes were greater than expected. Denitrification rates were high in saturated, fertilized soils, but low under all other conditions. Annual denitrification was calculated to be $14.0 \pm 3.6 \text{ kg N ha}^{-1} \text{ yr}^{-1}$, with 5% of the growing season accounting for >80% of the annual activity. Denitrification is thus an important means of removing reactive N in residential landscapes, but varies markedly in space, time, and with factors that affect soil saturation (texture, structure, compaction) and NO_3^- availability (fertilization). Rates of in situ N_2O flux were low; however, when recently fertilized soils saturated with water were incubated in the laboratory, we saw extraordinarily high rates of N_2O production for the first few hours of incubation, followed by rapid N_2O consumption later in the experiment. These findings indicate a lag time between accelerated N_2O production and counterbalancing increases in N_2O consumption; thus, we cannot yet conclude that lawns are an insignificant source of N_2O in our study area.

THE DEVELOPED LAND AREA of the United States has grown more than 400% in the past half century (Brown et al., 2005) and continues to grow rapidly (Goetz et al., 2004; Jantz et al., 2005; Brown et al., 2005). The most visible changes that accompany development are increased impervious surface area (e.g., buildings, roads, and sidewalks) and replacement of natural vegetation and agriculture fields with lawns (i.e., turfgrass). Impervious surfaces in the United States are estimated to cover an area nearly the size of Ohio ($112,610 \text{ km}^2$; Elvidge et al., 2004), and the area in lawns is estimated to be even larger ($163,812 \text{ km}^2$; Milesi et al., 2005).

The rapid increase in residential land area has raised concerns about pollution associated with fertilizers and pesticides, which are used for lawn establishment and maintenance (Morton et al., 1988; Gold et al., 1988, 1990; Petrovic, 1990; Milesi et al., 2005). Concern for coastal and freshwater ecosystems is particularly high, where nitrogen pollution has contributed to numerous water quality problems, including harmful algal blooms, decreased biodiversity, fisheries declines, and “dead zones” in areas such as the Gulf of Mexico and Chesapeake Bay (Carpenter et al., 1998; Kemp et al., 2005; Paerl et al., 2006; Dodds et al., 2008).

Residential systems receive higher N inputs than natural systems and may have a high capacity for N retention (Baker et al., 2001; Wollheim et al., 2005; Groffman et al., 2009). Lawns, which are the dominant vegetation cover in residential areas, may play an important role in this outcome. While lawns can have significant N losses, especially when overwatered and overfertilized (Morton et al., 1988; Petrovic, 1990; Qian et al., 2003), they also have considerable potential for organic matter accumulation and N retention (Gold et al., 1990; Qian and Follett, 2002; Kaye et al., 2005; Golubiewski, 2006; Raciti et al., 2008).

Through our studies of the Baltimore Long-Term Ecological Research (LTER) site, we have measured N inputs, outputs, and transformations in residential ecosystems with a particular focus on lawns. In a ^{15}N -tracer study we found that when a pulse of nitrate (NO_3^-) (simulating atmospheric deposition) was added to Baltimore area lawns, the NO_3^- was rapidly incorporated or retained in microbial biomass, fine roots, aboveground vegetation biomass, thatch, and soil organic matter (Raciti et al., 2008). Overall ^{15}N retention was similar, and possibly higher, than in

Copyright © 2011 by the American Society of Agronomy, Crop Science Society of America, and Soil Science Society of America. All rights reserved. No part of this periodical may be reproduced or transmitted in any form or by any means, electronic or mechanical, including photocopying, recording, or any information storage and retrieval system, without permission in writing from the publisher.

J. Environ. Qual. 40:1932–1940 (2011)

doi:10.2134/jeq2011.0107

Posted online 26 Aug. 2011.

Received 24 Mar. 2011.

*Corresponding author (raciti@bu.edu).

© ASA, CSSA, SSSA

5585 Guilford Rd., Madison, WI 53711 USA

S.M. Raciti and T.J. Fahey, Dep. of Natural Resources, Cornell Univ., Fernow Hall, Ithaca, NY 14853; A.J. Burgin, Dep. of Earth & Environmental Sciences, Wright State Univ., Dayton, OH 45435; P.M. Groffman and D.N. Lewis, Cary Institute of Ecosystem Studies, Box AB, Millbrook, NY 12545. S.M. Raciti, current address: Dep. of Geography and Environment, Boston Univ., 675 Commonwealth Ave., 4th Floor, Boston, MA 02215. Assigned to Associate Editor Philippe Vidon.

Abbreviations: BES, Baltimore Ecosystem Study; N-FARM, Nitrogen-Free Atmospheric Recirculation Method; PVC, polyvinyl chloride.

forested reference sites. In a second study, we evaluated soil carbon (C) and nitrogen (N) pools in 32 residential home lawns down to 1-m depth (Raciti et al., 2011a). We found evidence for rapid accumulation of C and N in residential soils, particularly in those with a history of agricultural land use. When we studied NO_3^- production and consumption in the same lawns (Raciti et al., 2011b), we found that net nitrification and exchangeable NO_3^- were significantly higher in residential soils than in forest soils, but these measures of NO_3^- production and availability were still notably low and comparable to deciduous forest stands in other studies (Lovett et al., 2004). Finally, while N inputs to our developed watersheds were high (Law et al., 2004), measured leaching losses were a relatively small fraction of total inputs (Groffman et al., 2009). Using information gathered from these experiments, we created a mass balance for our lawn-dominated residential systems (Raciti et al., 2011b); however, a major N flux was missing from our balance—gaseous losses of N_2 from denitrification—a process that has been notoriously difficult to measure in terrestrial ecosystems (Kulkarni et al., 2008).

Direct measurements of denitrification in terrestrial ecosystems have been limited by the difficulty of measuring N_2 fluxes (see review by Groffman et al., 2006a); however, past research suggests that denitrification may be an important pathway for N loss from turfgrass systems (e.g., Horgan et al. 2002a,b). Due to the difficulty in measuring N_2 fluxes, mass balance studies have generally estimated denitrification as the difference between recovered and unrecovered fertilizer N at the end of an experiment (Horgan et al., 2002a). For instance, Miltner and colleagues (1996) recovered 64 to 81% of ^{15}N -labeled urea in the spring and fall of their experiment, suggesting that gaseous losses might be responsible for incomplete recovery. Engelsjord and colleagues (2004) and Starr and DeRoo (1981) reported a similar range of recovery rates. Horgan and colleagues (2002a,b) attempted to account for gaseous losses by periodically measuring ^{15}N - N_2O and ^{15}N - N_2 fluxes from lawn and greenhouse plots, but overall recoveries (57–84%) were not much improved over previous mass balance studies. Despite less than complete recovery, Horgan and colleagues calculated sizeable N_2 and N_2O fluxes, particularly following a major rainfall event, confirming that denitrification was an important process in turfgrass systems.

In this study, we sought to complete the N budget for our residential system (Raciti et al., 2011b) using a combination of laboratory and field measurements to quantify denitrification. There is general consensus on the proximate controls of denitrification in terrestrial ecosystems (e.g., soil temperature, moisture, oxygen, organic matter, and available NO_3^-). As an anaerobic, primarily heterotrophic process, denitrification in terrestrial environments tends to occur in small areas (hot-spots) and during brief periods (hot moments) of activity when these controls converge (Parkin, 1987; McClain et al., 2003). In addition to influencing the total rate of denitrification, these factors influence the relative importance of the end products of denitrification, including N_2O , which is an important greenhouse gas (Kulkarni et al., 2008). Two problems have hindered attempts to estimate meaningful rates outside the laboratory: (i) an inability to measure direct N_2 fluxes due to high atmospheric concentrations of this gas, and (ii) an inability to mea-

sure and monitor soil oxygen (O_2) concentrations in the field (Groffman et al., 2006a). Using recent advances in instrumentation (Butterbach-Bahl et al., 2002; Burgin et al., 2010), we were able to overcome these problems to measure field-relevant rates of denitrification from lawns. Our objectives were to (i) measure soil O_2 conditions in home lawns; (ii) determine relationships between soil O_2 conditions, and N_2 and N_2O fluxes, in the laboratory and field (N_2O only); (iii) evaluate the role of soil moisture and nitrate levels on these relationships; and (iv) produce estimates of annual N gas fluxes to close the N budget of our residential ecosystem. At the outset of the study, we hypothesized that denitrification (as N_2 and N_2O) would occur on an episodic basis, with the highest rates coinciding with rainfall and after fertilizer application, when soil moisture and available NO_3^- would be high and soil oxygen concentrations would be low.

Materials and Methods

Site Description

The lawns used in this study were located in the Baltimore metropolitan area in association with the Baltimore Ecosystem Study (BES; <http://beslter.org>), a component of the U.S. National Science Foundation LTER network, and have been described in detail elsewhere (Groffman et al., 2009). This area has a temperate climate with warm, humid summers (1220 cooling degree days), cold winters (4720 heating degree days), and mean annual precipitation of approximately 1060 mm distributed relatively evenly throughout the year (NCDC, 2009).

The two lawns (referred to as UMBC1 and South Campus) were located on the grounds of University of Maryland Baltimore County, but were not turfgrass research plots. Campus lawns were chosen because of their known, long-term management regimes, similarities in land-use history (formerly agricultural), and their representativeness of typical residential lawns. Each lawn (or plot) contained a mixture of tall fescue (*Festuca arundinacea* Schreb.) and fine fescue (*Festuca* spp.). The UMBC1 lawn, which was less actively managed, also had significant white clover (*Trifolium repens* L.) cover. UMBC1 was fertilized each spring at a rate of 98 kg N ha⁻¹ using LESCO 14–14–14 (LESCO, Cleveland, OH), applied in two applications, approximately 2 wk apart. Mowing was done at 2- to 3-wk intervals (dependent on rainfall and subsequent growth) during the spring, summer, and fall seasons to a height of approximately 10 cm. South Campus was fertilized each spring and fall at 98 kg N ha⁻¹ (196 kg N ha⁻¹ yr⁻¹ total). Fertilizer was applied over two applications each season, approximately 2 wk apart, using LESCO 18–24–12 in the spring and LESCO 25–5–10 in the fall. The lawn was mowed weekly during the growing season to a height of approximately 10 cm. Neither lawn received irrigation, and clippings were left in place. Management of the lawns was similar to moderate (UMBC1) and moderately high maintenance (South Campus) home lawns in the study area. A lawn management survey by Law et al. (2004) found that lawn fertilizer inputs in the area ranged from zero to >300 kg N ha⁻¹ yr⁻¹ with a mean of 83.5 kg N ha⁻¹ yr⁻¹ in their suburban watershed (0.13-ha mean lot size). The soils at UMBC1 and South Campus have been classified as Joppa (loamy-skeletal, siliceous, semiactive, mesic Typic Hapludult) and Brandywine

(sandy-skeletal, mixed mesic Typic Dystrudept), respectively. Soil pH was between 5.7 and 6.0 for both lawns when measured at the beginning of the study. Atmospheric N deposition in the Baltimore metropolitan area is estimated at 11.2 kg N ha⁻¹ yr⁻¹ (Groffman et al., 2004).

Instrumentation

In early August of 2008, each lawn was instrumented with three Apogee diffusion-head soil O₂ sensors (SO-100 series, Apogee Instruments, Logan, UT) with the diffusion heads at 7-cm depth. The sensors in each lawn were connected to a Campbell Scientific CR-800 data logger (Campbell Scientific, Logan, UT), which was programmed to measure soil O₂ hourly over the course of approximately 1 yr (with monthly interruptions to recharge and replace batteries at each installation). Both lawns have been continually monitored for soil temperature and moisture as part of the BES network of long-term study sites (Groffman et al., 2009).

Sample Collection and Processing

Soil cores were collected to a depth of 10 cm using a 5-cm-diameter slide-hammer corer (AMS Equipment Corp., American Falls, ID) from 13 May and 4 June 2009. For each experiment (described below) an equal number of soil cores were collected from each of the two lawn areas (UMBC1 and South Campus), with the exception of the field fertilization experiment (Experiment 3), where all of the soil cores came from the UMBC1 site. Soil cores were collected randomly from each lawn area. All cores were transported to the laboratory in coolers and then stored at 4°C until they could be processed (2–4 d). A subset of soil cores from each time point were used to determine soil dry weight, percent moisture, and bulk density. Soil dry weight and percent moisture were determined by the change in mass of 10 g of field-moist soil after 48 h at 105°C. Bulk density (BD) was calculated as $BD = (\text{Total Dry Mass} - \text{Rock Mass}) / (\text{Total Volume} - \text{Rock Volume})$ (see Boone et al., 1999).

Direct Measurement of N₂, Nitrous Oxide, and Carbon Dioxide Fluxes from Intact Lawn Soil Cores

Denitrification measurements were made using the Nitrogen-Free Atmospheric Recirculation Method (N-FARM) system (Burgin et al., 2010), which is based on those built by Swerts et al. (1995) and Butterbach-Bahl et al. (2002). Intact soil cores were encased in 10 glass chambers (each acting as an independent incubator) with gastight lids and Swagelok fittings and valves connected to a gastight flow injection system built from Swagelok connections (Swagelok, Crawford Fitting Co., Solon, OH). These were mounted in-line with Shimadzu (Kyoto, Japan) GC8A gas chromatographs with thermal conductivity (to measure N₂, CO₂, and O₂) and electron capture (to measure N₂O) detectors. The glass chambers were submerged under water in a Plexiglas box. This served as a secondary test for leaks (which would be evident from water accumulation in the chambers) and further minimized potential gas diffusion around the chamber seals. Using this system, the ambient soil atmosphere could be replaced with an N₂-free helium (He) and O₂ atmosphere using a series of gentle vacuum-flush cycles (Butterbach-Bahl et al., 2002; Burgin et al., 2010). By

lowering N₂ concentration in the soil atmosphere (to below the 7.5 μmol mol⁻¹ detection limit of our thermal conductivity detector), we were able to measure small fluxes of N₂ gas exchange from intact soil cores. The system allowed us to vary soil O₂ concentrations by adjusting the He:O₂ ratio used during the vacuum-flush cycle. Regular system pressure checks, empty chambers (two blanks per incubation), and incubation of killed soil cores (autoclaved twice at 125°C for 3 h over the course of 3 d) were used to control for system leaks and residual degassing of N₂ from soil pores. All soil cores were incubated at 25°C and left completely intact (including plant roots and shoots) to minimize disturbance to the soil ecosystem.

Field Fluxes

In the field, soil:atmosphere fluxes of CO₂ and N₂O were measured using 29-cm (inner diameter) polyvinyl chloride (PVC) cylinder chambers with gas sampling ports (Bowden et al., 1991; Groffman et al., 2006b). Just before sampling, these chambers were mounted on PVC base rings installed to 5-cm depth and flush with the soil surface. These low-profile base rings allowed mowing to take place as usual between sampling intervals. At 0, 10, 20, and 30 min following placement of the chamber on the base, 9-mL gas samples were collected from gas sampling ports in the center of the chamber top by syringe. Samples were transferred to evacuated glass vials which were stored at room temperature before analysis by gas chromatography with electron capture (N₂O) or thermal conductivity (CO₂) detection.

Experiments

The first experiment tested the effects of O₂ concentration on N₂ and CO₂ fluxes at field moisture (17–20% volumetric soil moisture). Using intact soil cores, N₂ and CO₂ fluxes were measured at 20, 5, and 0% O₂ concentrations in the N-FARM. Each soil O₂ concentration was tested using fresh soils ($n = 4$) to avoid unknown effects that might arise from subjecting soil cores to prolonged incubation. Headspace N₂ and CO₂ concentrations were measured after 2, 6, and 24 h of incubation.

Our second experiment tested the effect of soil moisture and NO₃⁻ availability on N₂ and CO₂ fluxes at 0% O₂. Wet soil cores (>30% volumetric soil moisture, $n = 4$) were collected during a rain event and compared with dry soil cores (18% soil moisture, $n = 4$) collected before the rain event. Since similarly low N₂ fluxes were obtained for both conditions, a third incubation was conducted at 0% O₂ after injecting 10 mL of NO₃⁻ solution (approximately 100 mg N kg⁻¹ dry soil) into four wet soil cores. Headspace N₂, N₂O, and CO₂ concentrations were measured after 2, 6, and 24 h of incubation.

Our third experiment examined changes in soil N₂, N₂O, and CO₂ fluxes following a field fertilization experiment. On 13 May 2009, a 3- by 30-m section of the UMBC1 site was amended with LESCO 24-0-11 fertilizer at a rate of 49 kg N ha⁻¹ using a calibrated spreader. Soil cores were collected from the recently fertilized and “ambient” (i.e., control) areas of the plot. The first N-FARM incubation was performed at 20% O₂ using ambient ($n = 4$) and recently fertilized ($n = 4$) soil cores taken 24 h after fertilizer addition. The incubation was performed again at 2% soil O₂ using fresh cores. The experiment was repeated for cores taken 9 d after fertilizer addition. The fertilized area of UMBC1 included three gas rings that were used to measure N₂O and

CO₂ fluxes in the field. Field fluxes were measured before fertilizer application, and 24 h and 9 d after application (i.e., at the same time that soil cores were collected). All soil cores were incubated at field moisture (18–21% volumetric soil moisture).

We performed two more N-FARM incubations to test the effects of soil saturation (e.g., from a major rainstorm) on denitrification rates in fertilized soils. Field fertilized soil cores (collected 9 d postfertilization) were saturated by slowly dripping water onto them until no further water could be absorbed. These cores were incubated at 20 and 2% O₂ ($n = 8$ for each O₂ concentration).

Calculations and Statistics

Linear regression (accumulation of N₂, N₂O, or CO₂ vs. time) was used to calculate flux rates from N-FARM incubations. We used killed soil cores to correct for low levels of N₂ accumulation that might have originated from degassing from soil pores. Fluxes were expressed on a per-area basis by dividing by the ground surface area of each soil core. Field fluxes (using PVC cylinder chambers described earlier) were calculated from the linear rate of change in gas concentration, the chamber internal volume and soil surface area (Groffman et al., 2006b, 2009). Statistical differences between experimental treatments were tested using analysis of variance followed by Tukey's Honestly Significant Difference post hoc test. All statistical analyses were performed using SAS JMP version 8 statistical software (SAS Institute, 2009). For laboratory Experiments 1 and 2, an equal number of soil cores from each lawn were used on the grounds that the different management regimes would result in differences in N₂, N₂O, and CO₂ fluxes. We found no statistically significant differences between the two lawns across laboratory experiments, so the results for both lawns are grouped in the figures and statistics. Using a combination of laboratory and field measurements, we estimated annual N₂ fluxes in residential lawns (Table 1). Annual N₂ fluxes were estimated by apportioning flux rates from appropriate N-FARM incubations to the proportion of the growing season (231 d for Baltimore, MD [NCDC, 2009]) when field-measured soil moisture and O₂ conditions fell within that range.

Results

We measured in situ soil O₂ concentrations hourly for 1 yr and found that concentrations stayed within a relatively small range (Fig. 1A and 1B) near atmospheric concentrations (20%), occasionally dropping as low as 17%. A 3-wk period with two major storm systems (4.3 and 10.4 cm of precipitation) failed to cause a significant drop in bulk soil O₂ concentrations at our sites.

In laboratory incubations we tested for the effect of soil O₂ concentration on N₂ and CO₂ fluxes (at 20% field moisture), and found that CO₂ fluxes (i.e., respiration) decreased as O₂ concentrations were lowered ($p < 0.02$ for all differences). The

N₂ fluxes increased as O₂ decreased, but the differences were not statistically significant due to high variability within treatments (Fig. 2A and 2B). The N₂ fluxes were low across the O₂ concentration gradient, with means ranging from $4.26 \pm 3.91 \mu\text{g N m}^{-2} \text{ h}^{-1}$ at 20% O₂ to $28.9 \pm 17.0 \mu\text{g N m}^{-2} \text{ h}^{-1}$ at 0% O₂.

We used changes in the field moisture of lawn soils (collected before and during a rain event) to measure the influence of soil moisture on N₂ and CO₂ fluxes at 0% soil O₂. While wet soils (>30% field moisture) had higher arithmetic means than dry soils (18% field moisture), these differences were not statistically significant ($59.8 \pm 13.9 \mu\text{g N m}^{-2} \text{ h}^{-1}$ vs. $28.9 \pm 17.0 \mu\text{g N m}^{-2} \text{ h}^{-1}$ and $9.1 \pm 1.5 \text{ mg C m}^{-2} \text{ h}^{-1}$ vs. $6.0 \pm 1.8 \text{ mg C m}^{-2} \text{ h}^{-1}$; Fig. 3A and 3B). However, when a pulse of NO₃[−] was added to the wet soils incubated at 0% O₂, N₂ fluxes increased dramatically from $59.8 \pm 13.9 \mu\text{g N m}^{-2} \text{ h}^{-1}$ to $2216 \pm 509 \mu\text{g N m}^{-2} \text{ h}^{-1}$ ($p < 0.001$). Carbon dioxide fluxes from wet soils amended with NO₃[−] were significantly higher than CO₂ fluxes from dry soils, but not significantly higher than fluxes in unamended wet soil cores (Fig. 3B).

Fertilized and unfertilized ("ambient") soil cores collected 24 h after fertilization in the field (applied at 49 kg N ha^{−1}) did not show significantly different N₂ and CO₂ fluxes when incubated at 20 and 2% O₂ (Fig. 4A and 4B). When this experiment was repeated with soil cores collected 9 d after fertilization (Fig. 4A), we found that soils incubated at 2% O₂ had higher denitrification rates than those incubated at 20% soil O₂; however, differences between fertilized and ambient soil cores were not statistically different at a given O₂ concentration. We found a predictable pattern of lower CO₂ fluxes at lower O₂ concentrations (Fig. 4B), but no differences in CO₂ fluxes between ambient and fertilized soils at a given O₂ concentration. Soil moisture was approximately 20% at both sampling intervals.

When we saturated field-fertilized soil cores (to simulate soil conditions after a major storm event), we found tremendously high N₂ fluxes at both 20 and 2% soil O₂ (Fig. 5A). The N₂ fluxes in saturated soils were $4039 \pm 723 \mu\text{g N m}^{-2} \text{ h}^{-1}$ and $2959 \pm 254 \mu\text{g N m}^{-2} \text{ h}^{-1}$ at 20% and 2% O₂, respectively, compared to $53.3 \pm 30 \mu\text{g N m}^{-2} \text{ h}^{-1}$ and $243 \pm 98 \mu\text{g N m}^{-2} \text{ h}^{-1}$ in field-moist soils. Carbon dioxide fluxes were lower at 2% O₂ than at 20% O₂ at each moisture level (Fig. 5B).

Field-measured fluxes of N₂O and CO₂ showed no significant differences 24 h after fertilization (Fig. 6A and 6B), but after 9 d fluxes had increased significantly compared to prefertilization rates ($2.00 \pm 0.62 \mu\text{g N}_2\text{O-N m}^{-2} \text{ h}^{-1}$ and $43.4 \pm 5.5 \text{ mg C m}^{-2} \text{ h}^{-1}$ vs. $-0.07 \pm 0.33 \mu\text{g N m}^{-2} \text{ h}^{-1}$ and $23.0 \pm 3.2 \text{ mg C m}^{-2} \text{ h}^{-1}$).

While we were able to measure N₂O concentrations in the incubation chambers, calculating N₂O fluxes would have been misleading due to the rapid and dynamic swings in concentration observed between sampling intervals. For instance, when fertilized soils were saturated with water (or when an NO₃[−] solution

Table 1. Annual denitrification rates calculated for study lawns based on laboratory measured N₂ flux rates and field measured soil moisture conditions, University of Maryland Baltimore County.

Volumetric soil moisture	Proportion of growing season	N ₂ flux kg N ha ^{−1} yr ^{−1}	Notes
8–35%	0.95	2.95 ± 1.65	Fertilized lawns 18–20% O ₂ , pH = 5.7–6.0
35%+	0.05	223.9 ± 40.1	Rate for saturated soils
Annual N ₂ flux		14.0 ± 3.6	

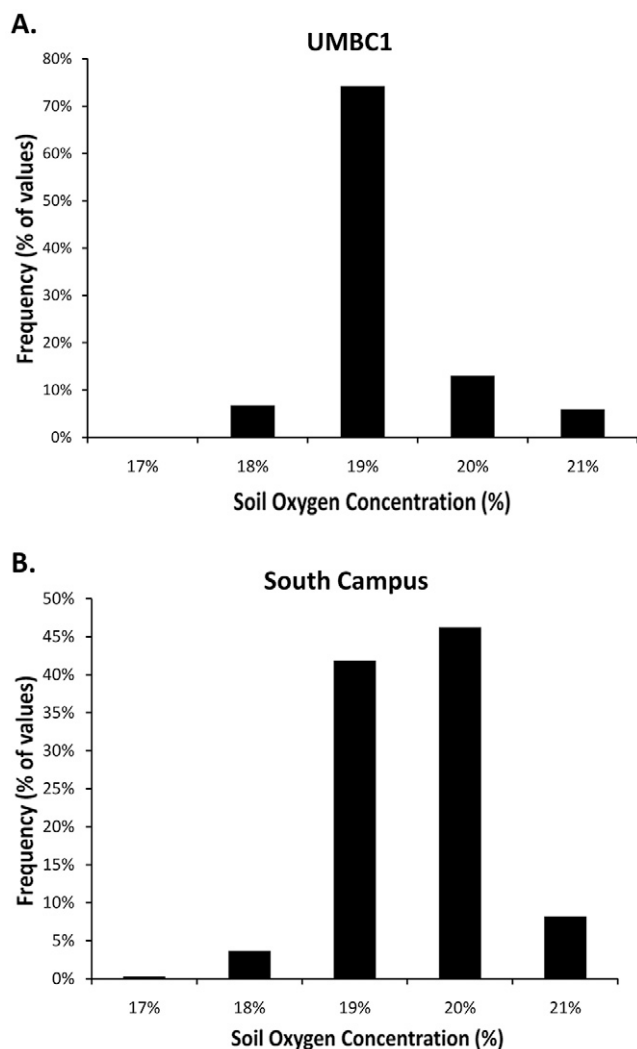


Fig. 1. Macroscale soil oxygen concentrations in two study lawns over a 1-yr time period: (A) UMBC1, and (B) South Campus, University of Maryland Baltimore County. Values for each lawn represent the mean of three sensors at 7-cm depth.

was added to wet soils), we saw extraordinarily high concentrations of N_2O for the first few hours of the incubation, followed by rapid N_2O consumption. In some cases, virtually all of the accumulated N_2O had been consumed before the next sampling interval (see Fig. 7 for examples from individual soil cores).

We estimated annual N_2 fluxes in residential lawns by apportioning flux rates from appropriate N-FARM incubations to the proportion of the growing season when field-measured soil moisture and O_2 conditions fell within that range (Table 1). We measured mean N_2 fluxes equivalent to $2.95 \pm 1.65 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ in UMBC1 and South Campus soils at 20% soil O_2 and 18 to 30% soil moisture. Based on soil O_2 and moisture measurements at the sites, similar conditions were found during 95% of the year, so a rate of $2.95 \pm 1.65 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ was assumed for this time period annually. Conversely, we measured very high N_2 flux rates (equivalent to $224 \pm 40.1 \text{ kg N ha}^{-1} \text{ yr}^{-1}$) at 20% O_2 in saturated, fertilized lawn soils. Based on maximum soil moisture readings at the sites (38%), we estimated that soils at 35% and greater soil water content would be approaching saturated conditions. Saturated conditions were experienced for 5% of the

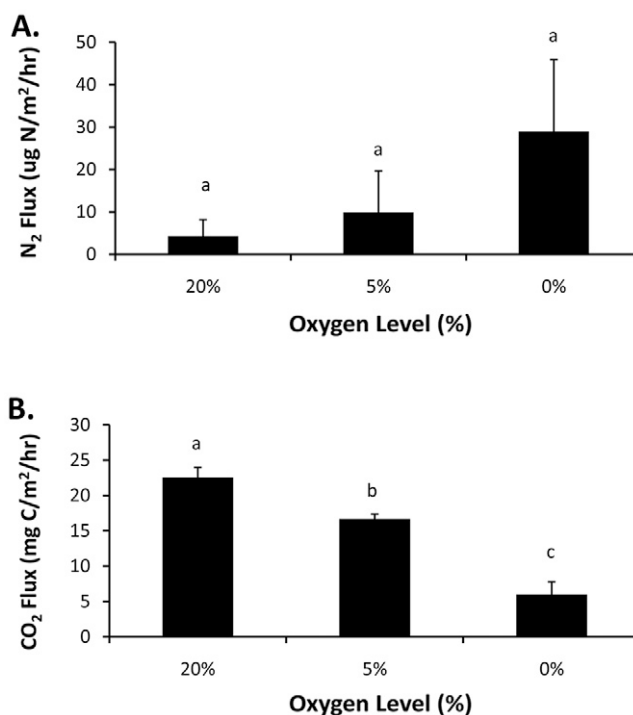


Fig. 2. (A) N_2 and (B) CO_2 fluxes from intact lawn soil cores at approximately 20% volumetric soil moisture across three soil oxygen concentrations (20, 5, and 0%), University of Maryland Baltimore County. Different letters indicate significant differences across treatments ($p < 0.05$, $n = 4$).

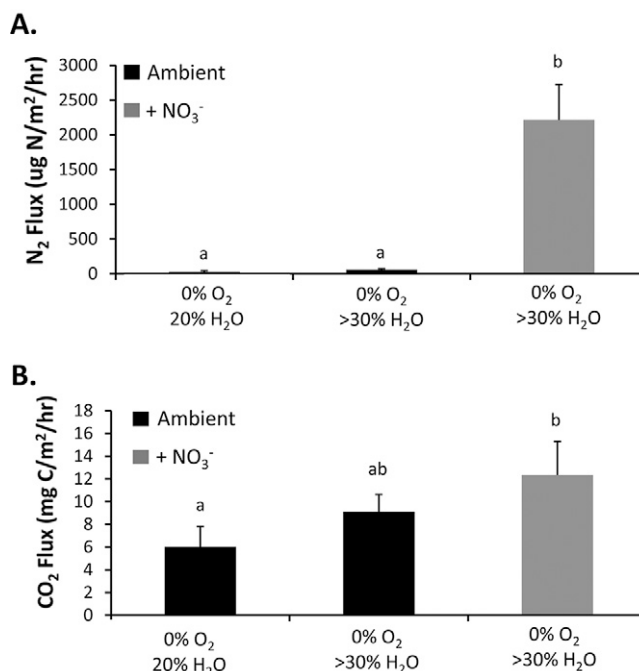


Fig. 3. (A) N_2 and (B) CO_2 fluxes from intact lawn soil cores at 0% oxygen, University of Maryland Baltimore County. Bars represent soil cores under ambient conditions before rainfall (approximately 20% volumetric soil moisture), ambient conditions after rainfall (>30% volumetric soil moisture), and injected with an NO_3^- solution after rainfall. Different letters indicate significant differences across treatments ($p < 0.05$, $n = 4$).

average growing season (based on 5 yr of soil moisture data). By this method, total annual denitrification was estimated at $14.0 \pm 3.6 \text{ kg N ha}^{-1} \text{ yr}^{-1}$.

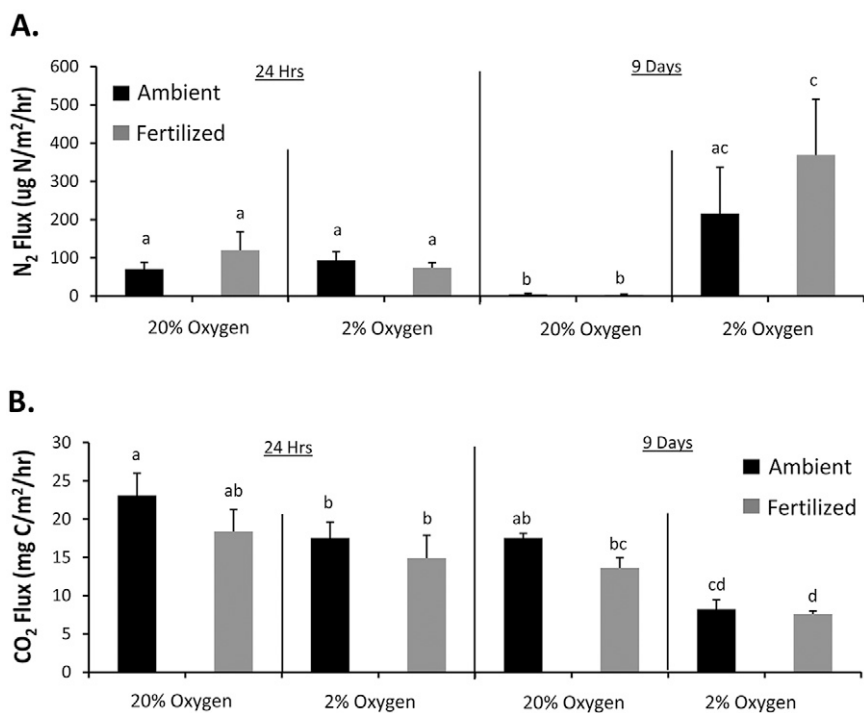


Fig. 4. (A) N₂ and (B) CO₂ fluxes at 20 and 2% oxygen for lawn soil cores collected 24 h and 9 d after a portion of each plot was fertilized in the field (Fertilized) or left untreated (Ambient), University of Maryland Baltimore County. Volumetric soil moisture was approximately 18 to 20%. Different letters indicate significant differences across all treatments and time periods ($p < 0.05$, $n = 4$).

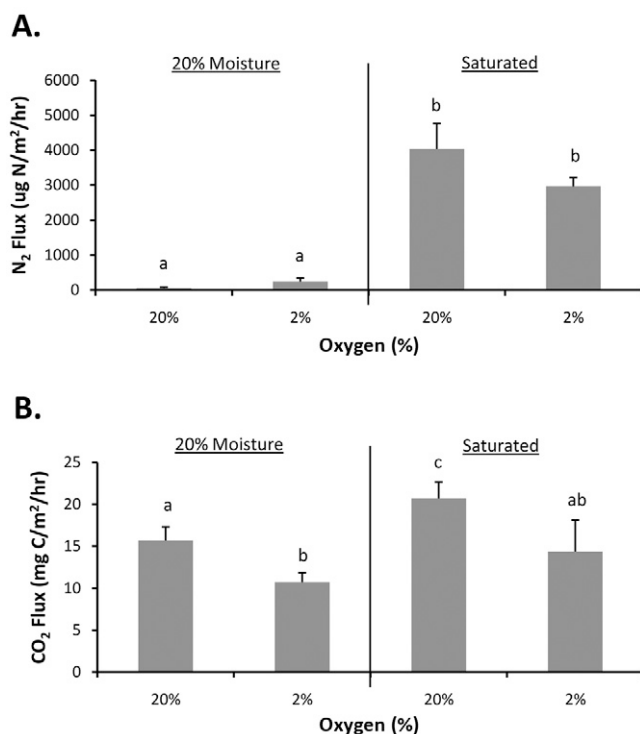


Fig. 5. (A) N₂ and (B) CO₂ fluxes at 20 and 2% oxygen for lawn soil cores collected 9 d after fertilization in the field, then saturated with water in the laboratory, University of Maryland Baltimore County. Different letters indicate significant differences across treatments ($p < 0.05$, $n = 8$).

Discussion

We hypothesized that (i) soil O₂, moisture, and available NO₃⁻ were the most important controls on denitrification in resi-

dential lawns; (ii) denitrification would vary temporally based on these controls; and (iii) N₂ fluxes would be low most of the time, with large pulses following fertilizer addition and precipitation events. While our results generally confirm these hypotheses, the threshold levels of soil O₂, moisture, and NO₃⁻ availability required to see significant rates of denitrification in lawns (relative to N inputs) were greater than expected. For instance, high rates of denitrification were possible when bulk soil O₂ concentrations were high, and for all oxygen concentrations, lawn soils had to be nearly saturated and recently fertilized to see significant N₂ fluxes. Soil oxygen concentrations were less dynamic (Fig. 1A and 1B), and N₂O fluxes even more dynamic than expected (Fig. 7).

Soil moisture has been linked to soil O₂ in a number of studies (Sextstone et al., 1985; Burgin et al., 2010), so we expected to measure significant drops in soil O₂ concentrations following large storm events; however, this was not the case. Soil O₂ concentrations (measured hourly by three diffusion-head O₂ sensors at each site) never dropped much below ambient atmospheric concentrations over our study period (Fig. 1A and 1B), which included several major storm events.

One storm system brought >10 cm of rainfall over a 3-d period (NCDC, 2009), yet soil O₂ concentrations did not change significantly at the sites. These findings suggest that the soils were well aerated and well drained. A lack of redoximorphic features in these soils further supports this conclusion.

We expected denitrification rates to be low when soil O₂ concentrations were high, but this was not always the case. Under wet or saturated conditions, soils amended with fertilizer or NO₃⁻ had high N₂ fluxes, even when soil O₂ concentrations were at 20% (4039 ± 723 μg N m⁻² h⁻¹; Fig. 5A). The Apogee sensors likely reflect O₂ levels in macropores, while denitrification likely occurs in anaerobic microsites associated with soil aggregates or particulate organic matter (Parkin, 1987). The fact that soil moisture appeared to be a stronger controller of denitrification rate than soil O₂ thus suggests that factors that influence the flux of O₂ from macropores to denitrifying microsites are critical controllers of denitrification rate. Given that these factors vary from site to site with soil texture, structure, compaction, and other variables, it is likely that relationships between soil O₂ and denitrification rates will vary in both time and space across the landscape. Another key factor affecting the levels of O₂ in soil microsites is bacterial and plant respiration. Lawn soils, which tend to have high soil organic matter, microbial biomass, and labile C and N, have correspondingly high rates of respiration (Qian and Follett, 2002; Raciti et al., 2008; Qian et al., 2010; Fig. 1B and 6B). Since respiration is the primary oxygen-consuming process in soils, we might expect anoxic microsites to form readily in lawns, especially under warm, moist conditions that favor high soil respiration.

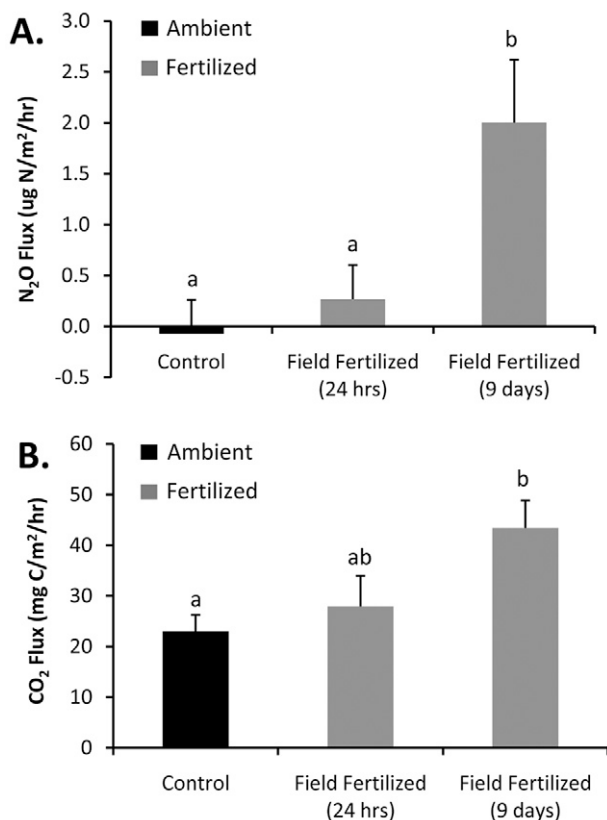


Fig. 6. (A) N₂O and (B) CO₂ fluxes from in situ gas sampling chambers in each of the lawn plots, University of Maryland Baltimore County. Different letters indicate significant differences across treatments ($p < 0.05$, $n = 6$).

The N₂O fluxes we measured in the field suggest that lawns are not a major source of N₂O emissions in our study system. When we fertilized part of the UMBC1 site, we expected to see an immediate and rapid increase in N₂O emissions (one of the intermediary gases in the denitrification process), but N₂O fluxes remained low after 24 h (Fig. 6A). We saw significantly higher N₂O fluxes 9 d after fertilizer application. These enhanced fluxes of N₂O may be the result of greater N availability 9 d after fertilizer application due to the use of a slow-release, sulfur-coated urea product (per standard management practices at the site). The extra time (9 d), combined with a light rainfall 3 d after fertilizer application, may have increased N availability in the soil. However, it should be stressed that even the enhanced N₂O fluxes ($<0.2 \text{ ng N cm}^{-2} \text{ h}^{-1}$) were lower than the mean rates measured in lawn and forest plots at BES (Groffman et al., 2009), which calls into question whether the low but significantly higher N₂O fluxes are a result of the experimental treatment or merely natural variation in the system.

Results from the N-FARM incubations suggest that soil moisture, NO₃⁻ availability, and the balance of N₂O production and consumption are key factors controlling N₂O emissions from lawns. Fluxes from wet, fertilizer-amended soils support the hypothesis that high soil moisture and NO₃⁻ availability can lead to high rates of N₂O production (Groffman et al., 2009), but these results also reveal a more complicated relationship between N₂O production and consumption. In fertilized cores, we sometimes saw extraordinarily high concentrations of N₂O followed by very low concentrations (near

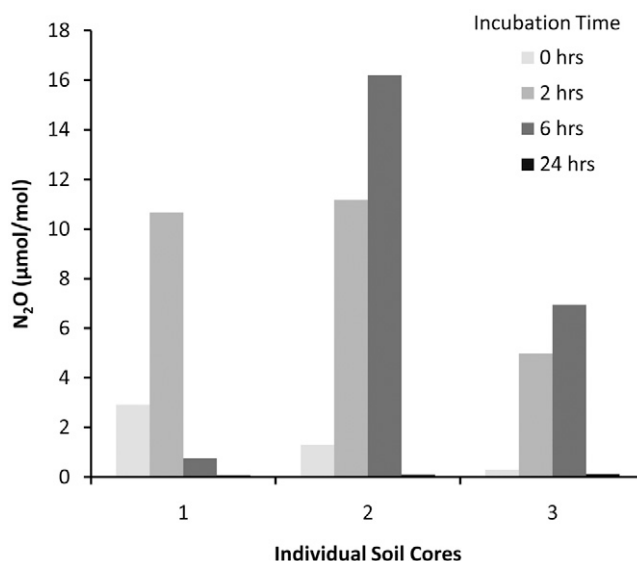


Fig. 7. Laboratory-measured N₂O concentrations in the headspace of individual soil sampling chambers at 0, 2, 6, and 24 h after the start of incubations for fertilized, saturated soil cores, University of Maryland Baltimore County.

the detection limit of our instrument) at the next sampling interval (Fig. 7). These findings indicate a lag time between accelerated N₂O production and counterbalancing increases in N₂O consumption, which may have important implications for N₂O fluxes from fertilized soils. Nitrous oxide is a by-product of both nitrification and denitrification, and its solubility and diffusivity control fluxes between the soil and the atmosphere (Heincke and Kaupenjohann, 1999). In the closed incubation system, N₂O could not escape and was generally consumed during the soil incubations. While soil saturation would inhibit N₂O diffusion to a significant degree (Heincke and Kaupenjohann, 1999), it is difficult to say how much N₂O would have been lost to the atmosphere. While long-term, in situ N₂O fluxes from BES lawns suggest that they are not major sources of N₂O emissions, these measurements are not keyed to fertilizer or rainfall events and may miss periods of high N₂O emissions (Groffman et al., 2009). Thus, like studies in the western United States (Kaye et al., 2004; Bijoor et al., 2008; Hall et al., 2008; Townsend-Small and Czimczik, 2010; Townsend-Small et al., 2011), we cannot yet conclude that lawns are an insignificant source of N₂O in the study area.

Closing the Nitrogen Budget

Using a combination of laboratory and field measurements, we calculated an annual denitrification rate of $14.0 \pm 3.6 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ for the lawns in this study (see the Calculations and Statistics section and Table 1). This calculation is based on long-term field and laboratory measurements of soil moisture and oxygen conditions, but does not account for other factors that might influence N₂ fluxes, such as temperature, seasonality (soil cores were collected during the spring), plant species, soil type, and climate variation. Further, while the sum of N inputs ($109.2 \text{ kg N ha}^{-1} \text{ yr}^{-1}$) and losses or sequestration ($111.1 \text{ kg N ha}^{-1} \text{ yr}^{-1}$) in our mass balance (Table 2) yields a nearly balanced N budget, there is significant uncertainty surrounding some of the numbers in the budget. The term with

Table 2. Nitrogen mass balance calculated for study lawns, University of Maryland Baltimore County.

Nitrogen flux	Residential	Forest	Source
N inputs			
Atmospheric deposition (kg N ha ⁻¹ yr ⁻¹)	11.2	11.2	Groffman et al., 2004
Fertilizer (kg N ha ⁻¹ yr ⁻¹)	98.0	0	This study (UMBC1 lawn plot)
Hydrologic N losses			
Leaching (kg N ha ⁻¹ yr ⁻¹)	14.1	4.4	Groffman et al., 2009
Gaseous N losses			
N ₂ O (kg N ha ⁻¹ yr ⁻¹)	<1	<1	Groffman et al., 2009
Denitrification as N ₂ (kg N ha ⁻¹ yr ⁻¹)	14.0	N/A	This study combined with long-term soil moisture records at Baltimore Ecosystem Study (http://beslter.org).
N sequestration			
N accumulation in soils (kg N ha ⁻¹ yr ⁻¹)	83.0	N/A	Residential sites with agricultural land-use history (Raciti et al., 2011a).

the largest uncertainty is soil N sequestration in residential lawn soils, which was estimated using a chronosequence of lawns in the study area (Raciti et al., 2011a) and may not be an accurate estimate for an individual lawn. Atmospheric N inputs may also be somewhat higher or lower than reported in our budget, since they reflect estimates from a USEPA Clean Air Status and Trends Network (CASTNET) site that is approximately 30 km to the south. Finally, hydrologic N losses are based on the long-term average for the site, but there is interannual variation in this rate (Groffman et al., 2009).

Despite these significant uncertainties, our calculations suggest that denitrification is an important part of the N budget for residential lawns (Table 2). The N₂ flux we calculated (14.0 ± 3.6 kg N ha⁻¹ yr⁻¹) was comparable in magnitude to atmospheric N deposition (11.2 kg N ha⁻¹ yr⁻¹), NO₃⁻ leaching losses (14.1 kg N ha⁻¹ yr⁻¹), and was equal to approximately 15% of mean fertilizer N inputs to home lawns in the study area (83.5 kg N ha⁻¹ yr⁻¹; Law et al., 2004). This rate is within the range of denitrification estimates obtained by Horgan et al. (2002a,b) using a ¹⁵N mass balance approach, though a significant fraction of the ¹⁵N tracer used in those studies could not be accounted for at the end of the experiments. Our annual flux calculation indicates that relatively short periods of saturation may be responsible for the majority of denitrification in our well-drained lawn soils. Lawns with limited drainage (or at low points in the landscape) could be hotspots for denitrification in residential areas. Our results suggest that denitrification is an important means of removing reactive N from the residential landscape, but varies markedly in space, time, and with factors that affect soil saturation (texture, structure, compaction) and NO₃⁻ availability (fertilization). While the annual denitrification rate we calculated was substantial, there is evidence that soil organic matter may be an even larger N sink in residential landscapes, particularly for sites with a history of agricultural land use before development (see Raciti et al., 2011a). Further work is required before the findings of this, or the aforementioned study, can be generalized to a wider range of residential lawns and soils.

Acknowledgments

This research was supported by the National Science Foundation Ecosystem Studies and LTER programs (Grants DEB 0614158, DEB-0444919, and DEB-9714835). The authors thank Dan Dillon for his tremendous help with field sampling and project planning. Additional thanks to Lisa Martel and Robin Schmidt for their help in the laboratory.

References

- Baker, L.A., D. Hope, Y. Xu, J. Edmonds, and L. Lauver. 2001. Nitrogen balance for the Central Arizona–Phoenix ecosystem. *Ecosystems* 4:582–602. doi:10.1007/s10021-001-0031-2
- Bijoor, N.S., C.I. Czimczik, D.E. Pataki, and S.A. Billings. 2008. Effects of temperature and fertilization on nitrogen cycling and community composition of an urban lawn. *Glob. Change Biol.* 14:2119–2131. doi:10.1111/j.1365-2486.2008.01617.x
- Boone, R.D., D.F. Grigal, R.A. Ahrens, P. Sollins, and D.E. Armstrong. 1999. Soil sampling, preparation, archiving, and quality control. p. 3–28. *In* G.P. Robertson et al. (ed.) *Standard soil methods for long-term ecological research*. Oxford Univ. Press, New York.
- Bowden, R.D., J.M. Melillo, P.A. Steudler, and J.D. Aber. 1991. Effects of nitrogen additions on annual nitrous oxide fluxes from temperate forest soils in the northeastern United States. *J. Geophys. Res. Atmos.* 96:9321–9328. doi:10.1029/91JD00151
- Brown, D.G., K.M. Johnson, T.R. Loveland, and D.M. Theobald. 2005. Rural land-use trends in the conterminous United States, 1950–2000. *Ecol. Appl.* 15:1851–1863. doi:10.1890/03-5220
- Burgin, A.J., P.M. Groffman, and D.N. Lewis. 2010. Factors regulating denitrification in a riparian wetland. *Soil Sci. Soc. Am. J.* 74:1826–1833. doi:10.2136/sssaj2009.0463
- Butterbach-Bahl, K., G. Willibald, and H. Papen. 2002. Soil core method for direct simultaneous determination of N₂ and N₂O emissions from forest soils. *Plant Soil* 240:105–116. doi:10.1023/A:1015870518723
- Carpenter, S.R., N.F. Caraco, D.L. Correll, R.W. Howarth, A.N. Sharples, and V.H. Smith. 1998. Nonpoint pollution of surface waters with phosphorus and nitrogen. *Ecol. Appl.* 8:559–568. doi:10.1890/1051-0761(1998)008[0559:NPOSWW]2.0.CO;2
- Dodds, W.K., W.W. Bouska, J.L. Eitzmann, T.J. Pilger, K.L. Pitts, A.J. Riley, J.T. Schloesser, and J. Darren. 2008. Eutrophication of U.S. freshwaters: Analysis of potential economic damages. *Environ. Sci. Technol.* 43:12–19.
- Elvidge, C.D., C. Milesi, J.B. Dietz, B.T. Tuttle, P.C. Sutton, R. Nemani, and J.E. Vogelmann. 2004. U.S. constructed area approaches the size of Ohio. *EOS Trans. Am. Geophys. Union* 85:233–234.
- Engelsjord, M.E., B.E. Branham, and B.P. Horgan. 2004. The fate of nitrogen-15 ammonium sulfate applied to Kentucky bluegrass and perennial ryegrass turfs. *Crop Sci.* 44:1341–1347. doi:10.2135/cropsci2004.1341
- Goetz, S.J., C.A. Jantz, S.D. Prince, A.J. Smith, R. Wright, and D. Varlyguin. 2004. Integrated analysis of ecosystem interactions with land use change: The Chesapeake Bay watershed. p. 263–275. *In* R.S. DeFries et al. (ed.) *Ecosystems and land use change*. Geophysical Monog. Ser., Am. Geophysical Union, Washington, DC.
- Gold, A.J., W.R. DeRagon, W.M. Sullivan, and J.L. Lemunyon. 1990. Nitrate-nitrogen losses to groundwater from rural and suburban land uses. *J. Soil Water Conserv.* 45:305–310.
- Gold, A.J., T.G. Morton, W.M. Sullivan, and J. McClory. 1988. Leaching of 2,4-D and dicamba from home lawns. *Water Air Soil Pollut.* 37:121–129. doi:10.1007/BF00226484
- Golubiewski, N.E. 2006. Urbanization increases grassland carbon pools: Effects of landscaping in Colorado's Front Range. *Ecol. Appl.* 16:555–571. doi:10.1890/1051-0761(2006)016[0555:UIGCPE]2.0.CO;2
- Groffman, P.M., M.A. Altabet, J.K. Böhlke, K. Butterbach-Bahl, M.B. David, M.K. Firestone, A.E. Giblin, T.M. Kana, L.P. Nielsen, and M.A. Voytek. 2006a. Methods for measuring denitrification: Diverse approaches to a difficult problem. *Ecol. Appl.* 16:2091–2122. doi:10.1890/1051-0761(2006)016[2091:MFMDDA]2.0.CO;2

- Groffman, P.M., J.P. Hardy, C.T. Drisoll, and T.J. Fahey. 2006b. Snow depth, soil freezing, and fluxes of carbon dioxide, nitrous oxide and methane in a northern hardwood forest. *Glob. Change Biol.* 12:1748–1760. doi:10.1111/j.1365-2486.2006.01194.x
- Groffman, P.M., N.L. Law, K.T. Belt, L.E. Band, and G.T. Fisher. 2004. Nitrogen fluxes and retention in urban watershed ecosystems. *Ecosystems* 7:393–403.
- Groffman, P.M., C.O. Williams, R.V. Pouyat, L.E. Band, and I.C. Yesilonis. 2009. Nitrate leaching and nitrous oxide flux in urban forests and grasslands. *J. Environ. Qual.* 38:1848–1860. doi:10.2134/jeq2008.0521
- Hall, S.J., D. Huber, and N.B. Grimm. 2008. Soil N₂O and NO emissions from an arid, urban ecosystem. *J. Geophys. Res.* 113:G01016. doi:10.1029/2007JG000523
- Heincke, M., and M. Kaupenjohann. 1999. Effects of soil solution on the dynamics of N₂O emissions: A review. *Nutr. Cycling Agroecosyst.* 55:133–157. doi:10.1023/A:1009842011599
- Horgan, B.P., B.E. Branham, and R.L. Mulvaney. 2002a. Mass balance of ¹⁵N applied to Kentucky bluegrass including direct measurement of denitrification. *Crop Sci.* 42:1595–1601. doi:10.2135/cropsci2002.1595
- Horgan, B.P., B.E. Branham, and R.L. Mulvaney. 2002b. Direct measurement of denitrification using ¹⁵N-labeled fertilizer applied to turfgrass. *Crop Sci.* 42:1602–1610. doi:10.2135/cropsci2002.1602
- Jantz, P., S. Goetz, and C. Jantz. 2005. Urbanization and the loss of resource lands in the Chesapeake Bay watershed. *Environ. Manage.* 36:808–825. doi:10.1007/s00267-004-0315-3
- Kaye, J.P., I.C. Burke, A.R. Mosier, and J.P. Guerschman. 2004. Methane and nitrous oxide fluxes from urban soils to the atmosphere. *Ecol. Appl.* 14:975–981. doi:10.1890/03-5115
- Kaye, J.P., R.L. McCulley, and I.C. Burke. 2005. Carbon fluxes, nitrogen cycling, and soil microbial communities in adjacent urban, native and agricultural ecosystems. *Glob. Change Biol.* 11:575–587. doi:10.1111/j.1365-2486.2005.00921.x
- Kemp, W.M., W.R. Boynton, J.E. Adolf, D.F. Boesch, W.C. Boicourt, G. Brush, J.C. Cornwell, T.R. Fisher, P.M. Glibert, J.D. Hagy, L.W. Harding, E.D. Houde, D.G. Kimmel, W.D. Miller, R.I.E. Newell, M.R. Roman, E.M. Smith, and J.C. Stevenson. 2005. Eutrophication of Chesapeake Bay: Historical trends and ecological interactions. *Mar. Ecol. Prog. Ser.* 303:1–29. doi:10.3354/meps303001
- Kulkarni, M.V., P.M. Groffman, and J.B. Yavitt. 2008. Solving the global nitrogen problem: It's a gas! *Front. Ecol. Environ.* 6:199–206. doi:10.1890/060163
- Law, N.L., L.E. Band, and J.M. Grove. 2004. Nitrogen input from residential lawn care practices in suburban watersheds in Baltimore County, MD. *J. Environ. Manage.* 47:737–755.
- Lovett, G.M., K.C. Weathers, M.A. Arthur, and J.C. Schultz. 2004. Nitrogen cycling in a northern hardwood forest: Do species matter? *Biogeochemistry* 67:289–308. doi:10.1023/B:BIOG.0000015786.65466.f5
- McClain, M.E., E.W. Boyer, C.L. Dent, S.E. Gergel, N.B. Grimm, P.M. Groffman, S.C. Hart, J.W. Harvey, C.A. Johnston, E. Mayorga, W.H. McDowell, and G. Pinay. 2003. Biogeochemical hot spots and hot moments at the interface of terrestrial and aquatic ecosystems. *Ecosystems* 6:301–312. doi:10.1007/s10021-003-0161-9
- Milesi, C., S.W. Running, C.D. Elvidge, J.B. Dietz, B.T. Tuttle, and R.R. Nemani. 2005. Mapping and modeling the biogeochemical cycling of turf grasses in the United States. *Environ. Manage.* 36:426–438. doi:10.1007/s00267-004-0316-2
- Miltner, E.D., B.E. Branham, E.A. Paul, and P.E. Rieke. 1996. Leaching and mass balance of ¹⁵N-labeled urea applied to a Kentucky bluegrass turf. *Crop Sci.* 36:1427–1433. doi:10.2135/cropsci1996.0011183X003600060001x
- Morton, T.G., A.J. Gold, and W.M. Sullivan. 1988. Influence of overwatering and fertilization on nitrogen losses from home lawns. *J. Environ. Qual.* 17:124–130. doi:10.2134/jeq1988.00472425001700010019x
- National Climatic Data Center (NCDC). 2009. Online climate data directory. Available at <http://lwf.ncdc.noaa.gov/oa/climate/climatedata.html> (verified 19 July 2011). Natl. Climatic Data Cent., NOAA, Asheville, NC.
- Paerl, H.W., L.M. Valdes, B.L. Peierls, J.E. Adolf, and L.W. Harding, Jr. 2006. Anthropogenic and climatic influences on the eutrophication of large estuarine ecosystems. *Limnol. Oceanogr.* 51:448–462. doi:10.4319/lo.2006.51.1_part_2.0448
- Parkin, T.B. 1987. Soil microsites as a source of denitrification variability. *Soil Sci. Soc. Am. J.* 51:1194–1199. doi:10.2136/sssaj1987.03615995005100050019x
- Petrovic, A.M. 1990. The fate of nitrogenous fertilizers applied to turfgrass. *J. Environ. Qual.* 19:1–14. doi:10.2134/jeq1990.00472425001900010001x
- Qian, Y.L., W. Bandaranayake, W.J. Parton, B. Mecham, M.A. Harivandi, and A.R. Mosier. 2003. Long-term effects of clipping and nitrogen management in turfgrass on soil organic carbon and nitrogen dynamics: The CENTURY model simulation. *J. Environ. Qual.* 32:1694–1700. doi:10.2134/jeq2003.1694
- Qian, Y.L., and R.F. Follett. 2002. Assessing soil carbon sequestration in turfgrass systems using long-term soil testing data. *Agron. J.* 94:930–935. doi:10.2134/agronj2002.0930
- Qian, Y., R.F. Follett, and J.M. Kimble. 2010. Soil organic carbon input from urban turfgrasses. *Soil Sci. Soc. Am. J.* 74:366–371. doi:10.2136/sssaj2009.0075
- Raciti, S.M., P.M. Groffman, and T.J. Fahey. 2008. Nitrogen retention in urban lawns and forests. *Ecol. Appl.* 18:1615–1626. doi:10.1890/07-1062.1
- Raciti, S.M., P.M. Groffman, J.C. Jenkins, R.V. Pouyat, T.J. Fahey, S.T.A. Pickett, and M.L. Cadenasso. 2011a. Accumulation of carbon and nitrogen in residential soils with different land use histories. *Ecosystems* 14:287–297. doi:10.1007/s10021-010-9409-3
- Raciti, S.M., P.M. Groffman, J.C. Jenkins, R.V. Pouyat, T.J. Fahey, S.T.A. Pickett, and M.L. Cadenasso. 2011b. Nitrate production and availability in residential soils. *Ecol. Appl.* (In press). doi:10.1890/10-2009.1
- SAS Institute. 2009. JMP Version 8, 1989–2009. SAS Inst., Cary, NC.
- Sextstone, A.J., N.P. Revsbech, T.B. Parkin, and J.M. Tiedge. 1985. Direct measurement of oxygen profiles and denitrification rates in soil aggregates. *Soil Sci. Soc. Am. J.* 49:645–651. doi:10.2136/sssaj1985.03615995004900030024x
- Starr, J.L., and H.C. DeRoo. 1981. The fate of nitrogen fertilizer applied to turfgrass. *Crop Sci.* 21:531–536. doi:10.2135/cropsci1981.0011183X002100040014x
- Swerts, M., G. Uytterhoeven, R. Merckx, and K. Vlassak. 1995. Semicontinuous measurement of soil atmosphere gases with gas flow soil core method. *Soil Sci. Soc. Am. J.* 59:1336–1342. doi:10.2136/sssaj1995.03615995005900050020x
- Townsend-Small, A., and C.I. Czimczik. 2010. Carbon sequestration and greenhouse gas emissions in urban turf. *Geophys. Res. Lett.* 37:L02707. doi:10.1029/2009GL041675
- Townsend-Small, A., D.E. Pataki, C.I. Czimczik, and S.C. Tyler. 2011. Nitrous oxide emissions and isotopic composition in urban and agricultural systems in southern California. *J. Geophys. Res.* 116:G01013. doi:10.1029/2010JG001494
- Wollheim, W.M., B.A. Pellerin, C.J. Vorosmarty, and C.S. Hopkinson. 2005. N retention in urbanizing headwater catchments. *Ecosystems* 8:871–884. doi:10.1007/s10021-005-0178-3